

# Measuring Radon in Exhaled Breath

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A photograph of a rural landscape. In the foreground, there is a dirt road with a concrete curb. The ground is dry and sandy, with sparse, dry grass. In the middle ground, there is a small settlement with several buildings and a large, dark tree. The background shows rolling hills and a cloudy sky.

## The Problem:

How to assess lifetime radiation exposure for people living near or working in uranium mines?

This is a significant problem because there are both legacy sites – abandoned or remediated – and active uranium mining still in many parts of the world, including Africa

(pictured)



# Possible Approaches

**Urine samples** – they represent short-term exposure, not lifetime, and uranium is not the main radiation source in ore.

**Modeling exposure** - based on monitoring and sampling – difficult to extrapolate back over a lifetime.

**Whole body monitoring** - possible for lifetime measuring gamma radiation only, but high cost of non-portable machine.

**Radon in exhaled breath** - in our opinion the best indicator for assessing lifetime exposure.

# **Studies of uranium ore exposure have not yet used radon in exhaled breath:**

**NIH\* and CDC\*\* funded studies of uranium exposure measured urinary levels of uranium, and Rn exposure was estimated based on the situation in the U mines ...**

\* National Institutes of Health (USA) [www.nih.gov](http://www.nih.gov)

\*\* Centers for Disease Control and Prevention(USA)

[www.cdc.gov/niosh/pgms/worknotify/uranium.html](http://www.cdc.gov/niosh/pgms/worknotify/uranium.html)

**URANIUM 238 (U238)  
RADIOACTIVE DECAY**

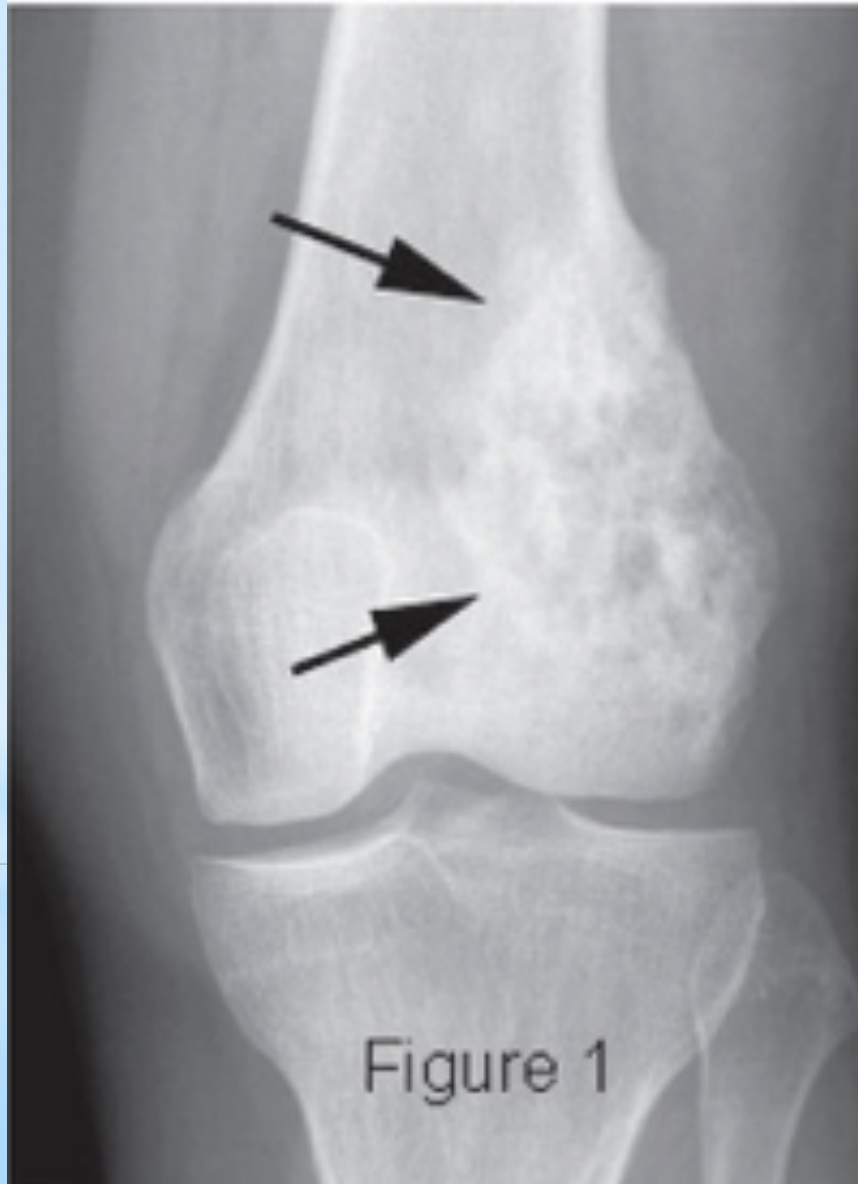
type of radiation	nuclide	half-life
	uranium—238	$4.5 \times 10^9$ years
$\alpha$	↓ thorium—234	24.5 days
$\beta$	↓ protactinium—234	1.14 minutes
$\beta$	↓ uranium—234	$2.33 \times 10^5$ years
$\alpha$	↓ thorium—230	$8.3 \times 10^4$ years
$\alpha$	↓ radium—226	1590 years
$\alpha$	↓ radon—222	3.825 days
$\alpha$	↓ polonium—218	3.05 minutes
$\alpha$	↓ lead—214	26.8 minutes
$\beta$	↓ bismuth—214	19.7 minutes
$\beta$	↓ polonium—214	$1.5 \times 10^{-4}$ seconds
$\alpha$	↓ lead—210	22 years
$\beta$	↓ bismuth—210	5 days
$\beta$	↓ polonium—210	140 days
$\alpha$	↓ lead—206	stable

Radium deposits in bone with long biological and radiological half lives

(3.825 d = 91.8 hrs )

Radon daughter isotopes

Radium deposited in



The method of **radon in exhaled breath** has been used with uranium miners who had **very high** exposures

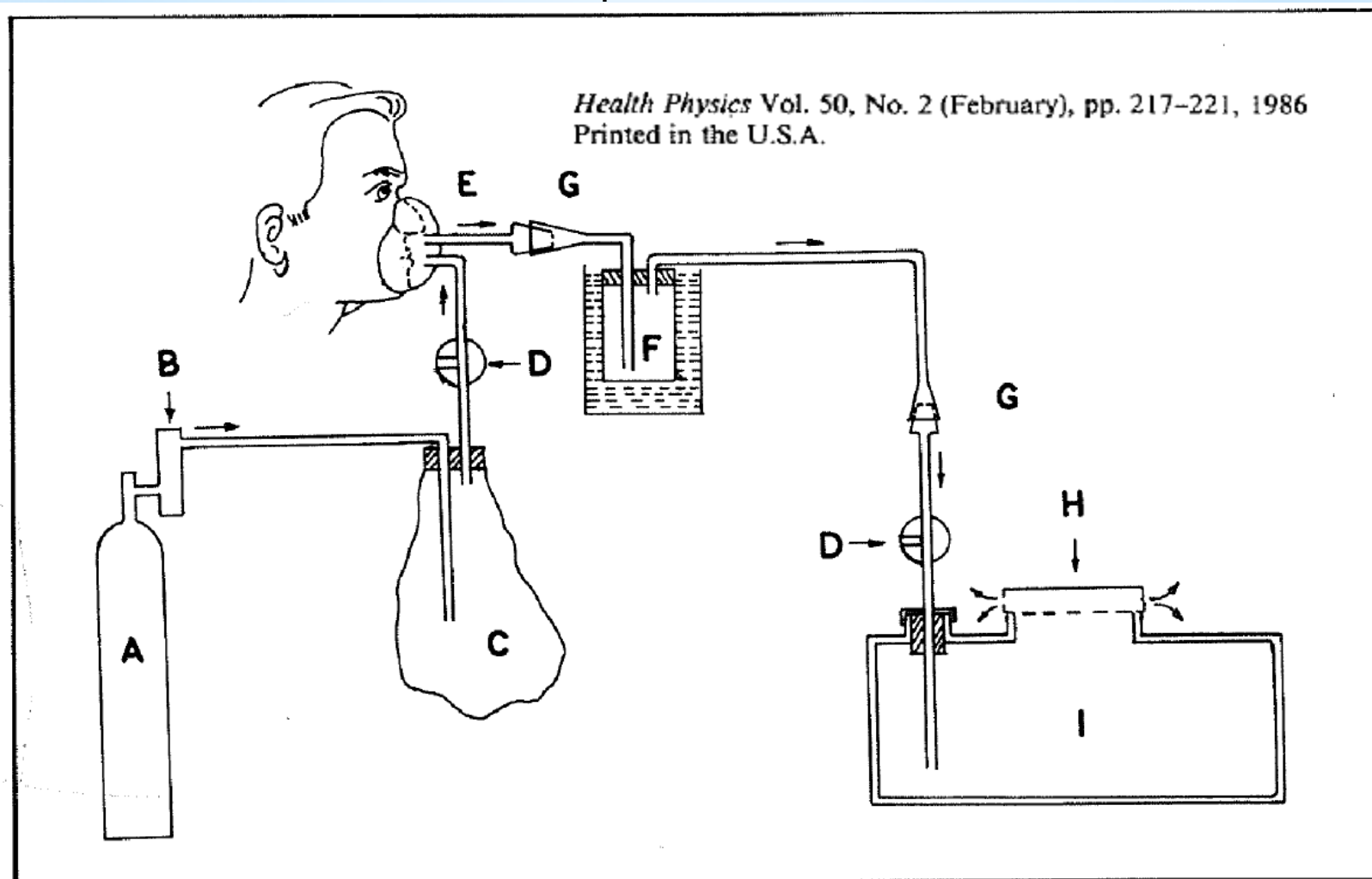


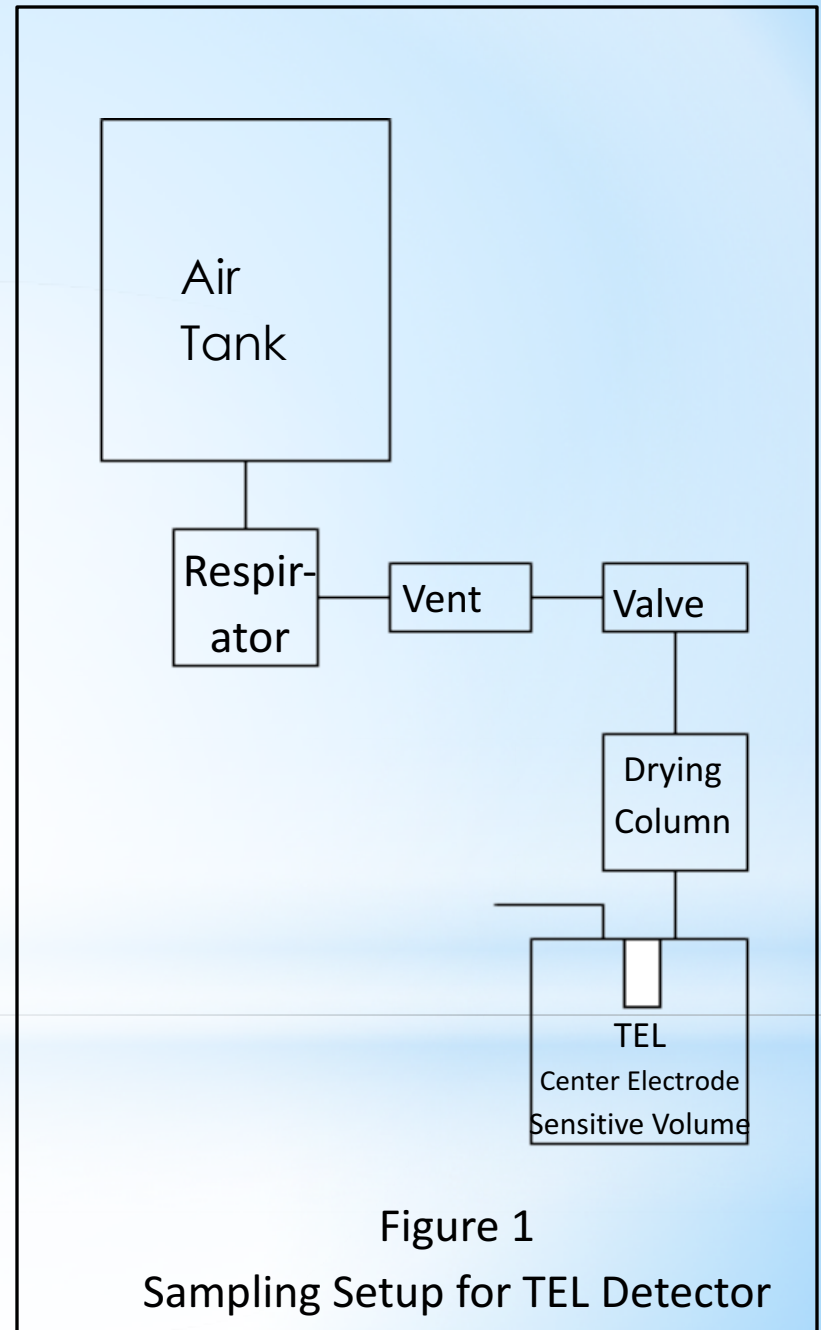
FIG. 1. Experimental setup for measurement of Rn in breath. A = medical oxygen cylinder, B = flow meter, C = plastic bag, D = three-way stopper, E = respirator, F = moisture trap, G = coupling, H = hair hygrometer, I = Rn collection chamber.



# Radiation detectors: A version is available from Pylon\*.

But sensitivity is still too low to  
measure low-level exposures.

\* <http://pylonelectronics-radon.com/>



The **Pylon Trace Level Radon Detector (TEL)** is the only commercially available radon detector with the **sensitivity of detection required** for measurements of radon in exhaled breath.

The high sensitivity of the TEL is due to its **large volume and low background noise level**.

But the **sensitivity should be further increased** by increasing the volume of the collection chamber. This would allow more rapid measuring individuals, thus facilitating epidemiological studies.

# Estimating $^{226}\text{Ra}$ Radium in the body from exhaled radon

Srivastava et al. (1986) note that the body content of  $^{226}\text{Ra}$  can be estimated from the amount of exhaled radon by the relationship:

<b>Q</b>	=	quantity of $^{226}\text{Ra}$ present in the body (pCi)
<b>I</b>	=	breathing rate of the subject (L/h)
<b><math>\lambda_{\text{Rn}}</math></b>	=	decay constant of $^{222}\text{Rn}$ (per h)
<b><math>C_{\text{Rn}}</math></b>	=	concentration of $^{222}\text{Rn}$ in the breath sample (pCi/L)
<b>f</b>	=	release fraction for $^{222}\text{Rn}$

# Estimating the release fraction f for $^{222}\text{Rn}$

**Srivastava *et al.* (1986) cited a value of 0.84 for f. This value is applicable to miners with high exposure.**

**Others (Dolan 1989) instead quoted a value for f of 0.63 in radium dial workers with long-term body burdens of  $^{226}\text{Ra}$  (Toohey *et al.* 1983). In these subjects, both  $^{226}\text{Ra}$  body content and  $^{222}\text{Rn}$  exhalation rates were measured at the same time, which lends additional credibility to the values.**

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Srivastava *et al.* Radium-226 body burden in U miners by measurement of radon in exhaled breath. *Health Phys.* 50:217-21 (1986)

Dolan, B. Memorandum to P. Groer "Relationship between breath radon measurements and skeletal radium burdens". Oak Ridge Associated Universities, September 1 (1989)

Toohey RE, Keane AT, Rundo J. Measurement techniques for radium and the actinides in man at the center for human radiobiology. *Health Phys.* 44 Suppl. 1:323-41 (1993)

Assuming a value of 0.63 and a standard breathing rate of 1.2 m<sup>3</sup>/h [“light work” from ICRP publication 66 (ICRP 2004)], **a value of 1 pCi/L of radon in the breath corresponds to approximately 0.25 μCi of <sup>226</sup>Ra in the body.**” (Bloom 2005).

This can be calculated by using the values above and equation (1) as shown below:

$$Q = \frac{1 \text{ pCi}}{L} \frac{1200L}{h} \frac{91.8h}{\ln(2)} \frac{1}{0.63} = 252,000 \text{ pCi} = 0.252 \text{ μCi}$$

Using a modified TEL detector the **lowest detection level is given with 0.008 pCi/L.**

# The lowest detection level of Rn

Using the manufacturer's suggested values of breathing rate and an  $f = 0.63$ , the following is obtained:

$$Q = (0.008 \text{ pCi/L})(1200 \text{ L/h})(91.8 \text{ h}/\ln(2))(1/0.63) = 2,100 \text{ pCi.}$$

**We estimate that a person exposed to background levels of radium might have a concentration between 0.016-0.023 pCi/g wet weight. Most of the radium is in bone; an adult weighing 60 kg has a skeleton that weighs about 12 kg or 12'000 g.  $12000 \text{ g} \times 0.02 \text{ pCi/g} = 240 \text{ pCi}$ .**

**In contrast, uranium miner body burdens of radium ranged from 3'500-56'700 pCi (Srivastava *et al.*, 1986).**

**We expect exposed community residents to fall in between these values and to mostly be within measurement range of the modified Pylon instrument.**

*Srivastava et al.* Radium-226 body burden in U miners by measurement of radon in exhaled breath. *Health Phys.* 50:217-21 (1986)

# Estimating Dose to Red Marrow

**Spiers *et al.* (1983) calculated the dose to the red marrow from  $^{226}\text{Ra}$  and its retained daughter isotopes**

$$\text{Marrow Dose} = 0.05 * 5.054 \left( \frac{A * E}{\text{BodyMass} * 0.1} \right) Y \text{ mGy}$$

Where A = whole-body Ra-226 content in Bq (=disintegrations/s) (from breath Rn-222)  
E = alpha particle energy in MeV absorbed in the skeleton per disintegration  
BodyMass = total body mass of the individual  
Y = age of subject or time spent in study area in years

**E can be calculated based on alpha particle energies of 4.77 MeV for  $^{226}\text{Ra}$ , 5.49 MeV for  $^{222}\text{Rn}$ , 6.00 MeV for  $^{218}\text{Po}$ , 7.69 MeV for  $^{214}\text{Po}$  and 5.30 MeV for  $^{210}\text{Po}$ . For environmental  $^{226}\text{Ra}$ , 30% retention is commonly assumed for its main decay products  $^{210}\text{Po}$  and  $^{210}\text{Pb}$  (Lloyd, 1991). This gives 11.32 MeV of alpha-particle energy per  $^{226}\text{Ra}$  decay.**

Spiers FW, Lucas HF, Rundo J, Anast GA. Leukaemia incidence in the U.S. dial workers. *Health Phys.* 44 Suppl. 1:65-72 (1993)

Lloyd RD & Bruenger FW. Rn:Ra ratios in bone of beagles injected with  $^{226}\text{Ra}$ . *Health Phys.* 60:567-68 (1991)